# Design, Synthesis, and Structure-Activity Relationship Studies of Novel 1-[(1-Acyl-4-piperidyl)methyl]-1*H*-2-methylimidazo[4,5-c]pyridine Derivatives as Potent, Orally Active Platelet-Activating Factor Antagonists

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Replacement of the polar head of our previous series of 1-acyl-4-[(2-methyl-3-pyridyl)cyanomethyl]piperazines with a 2-methylimidazo[4,5-c]pyridine group has led to the identification of a new series of 1-[(1-acyl-4-piperidyl)methyl]-1*H*-2-methylimidazo[4,5-c]pyridine derivatives as potent, orally active platelet-activating factor (PAF) antagonists. On the basis of the general structure—activity relationship trends found for the acyl substituent in our earlier series, five groups of compounds were tested, diaryl- or alkylarylpropanoyl derivatives, their 3-hydroxy-substituted analogues, and urea, carbamate and amino acid derivatives. The optimal compound 19 (UR-12670), bearing the 3,3-diphenylpropanoyl moiety, exhibited very high in vitro and in vivo potency (IC<sub>50</sub> = 0.0076  $\mu$ M for the in vitro PAF-induced platelet aggregation assay,  $ID_{50} = 0.0086$  mg/kg for the in vivo PAF-induced hypotension test in normotensive rats, and  $ID_{50} = 0.092$  mg/kg po and 0.0008 mg/kg iv for the PAF-induced mortality test in mice). Compound 19 also showed long duration of activity. It gave 100% protection against PAFinduced mortality in mice 7 h after iv administration of a single dose of 1 mg/kg and also provided 100% inhibition of PAF-induced aggregation in dog whole blood 6 h after iv administration of the same dose. The lead structure 19 has been selected for in-depth pharmacological evaluation.

## Introduction

Earlier work in our laboratory led to the discovery of (pyridylcyanomethyl)piperazines and -piperidines of general formula I and II as a new class of potent platelet-activating factor (PAF) antagonists and to the selection of UR-12460 (1) and UR-12519 (2) for development (see Table 1).1 These compounds, which have demonstrated to be fairly stable as solids, were found to suffer a relatively rapid hydrolysis of the (cyanomethyl)amino group in aqueous solution. This fact together with a short duration of action in vivo led us to discontinue them and initiate a search for a surrogate polar head within the same skeleton.

A structural feature of many PAF antagonists is the presence of an sp<sup>2</sup> nitrogen at a given distance from and orientation to an amide or other isosteric groups.2 We demonstrated that piperazinic derivatives I and their carba analogues II featured essentially the same activity in in vitro and iv in vivo tests (see 1 vs 3 in Table 1), showing that the piperazinic nitrogen serves merely as a framework. The presence of the cyano group, on the other hand, was essential for activity, suggesting that in addition to the requisite named above another feature may be important in the interaction of this type of compound with the PAF receptor, namely, a third coordination center close to the pyridine  $sp^{\check{z}}$  nitrogen. Inspection of the molecular model for the imidazo[4,5c|pyridine moiety, which is present in different PAF antagonists,<sup>3</sup> indicates that the two more distant nitrogens might also fulfill the latter requisite. Accordingly, structures like III resulting from the replacement of the (cyanomethyl)pyridyl moiety of II with 2-methylimidazo-[4,5-c] pyridine appear to be interesting candidates. This hypothesis is illustrated in Figure 1, which shows the MM2-optimized modeled structures<sup>4</sup> for (cyanomethyl)pyridine II ( $R = CH_3$ ) and the proposed imidazopyridine series III ( $R = CH_3$ ) overlaid with the three indicated

**Figure 1.** Overlay of (cyanomethyl)pyridine II (R = Me) and imidazopyridine III (R = Me).

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coordination centers of both molecules superimposed. In order to test our hypothesis, compounds 4-7, with the heterocyclic head linked directly to the piperidine ring or through a methylene bridge, were prepared (see Table 1). The activities of the longer compounds 4 and **5** were very close to those of their analogues **1–3**, and preliminary chemical stability studies indicated that the new skeleton was free of hydrolysis problems. With this result on hand, we proceeded to the fine optimization of the acyl radical following the trends found in previous series. The present report provides an account of the synthesis and pharmacological evaluation of 1-[(1-acyl-4-piperidyl)methyl]-1H-2-methylimidazo[4,5-c]pyridine derivatives, from which a new lead, 19, has been selected.

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Table 1. Comparison of (Cyanomethyl)pyridines with Imidazopyridines

comp	R	PAF-induced platelet aggregation IC50, <sup>a</sup> μΜ	PAF-induced hypotension ID50, <sup>b</sup> mg/kg iv	PAF-induced mortality ID50, <sup>c</sup> mg/kg iv	PAF-induced mortality ID50, <sup>d</sup> mg/kg po	mp, <sup>o</sup> C (cryst solvent) <sup>e</sup>	formula <sup>f</sup>
1	Ph CN N	0.040 (0.030-0.053)	0.021 (0.015-0.031)	0.009 (0.008-0.011)	0.30 (0.16-0.54)		
2	MeO N N N N N N N N N N N N N N N N N N N	0.041 (0.031-0.055)	0.015 (0.011-0.021)	0.0055 (0.0009-0.031)	0.044 (0.023-0.091)		
3	Ph N N N	0.024 (0.015-0.039)	0.030 (0.019-0.047)		2.6 (0.72-9.3)		
4	Ph Ph N N N N N N N	0.013 (0.009-0.017)	0.067 (0.036-0.13)	0.0041 (0.0014-0.012)	0.3-1	77-79	C <sub>28</sub> H <sub>31</sub> N <sub>5</sub> O.1/2H <sub>2</sub> O
5	MeO H N N N N N N N N N N N N N N N N N N	0.012 (0.009-0.017)	0.035 (0.027-0.045)	0.0027 (0.0009-0.0078)	0.03-0.1	97-100	C <sub>24</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub> .1/2H <sub>2</sub>
6	Ph N N N N	0.10 (0.05-0.2)	0.41 (0.31-0.55)		>3	95-100	C <sub>27</sub> H <sub>28</sub> N <sub>4</sub> O.1/4 H <sub>2</sub>
7	MeO Ph O	1.30 (1.0-1.75)	0.56 (0.27-1.14)			102-105	C <sub>23</sub> H <sub>27</sub> N <sub>5</sub> O <sub>3</sub> .H <sub>2</sub> O

<sup>a</sup> Concentration required to inhibit PAF-induced maximum aggregation by 50%. Parentheses contain 95% confidence limits. <sup>b</sup> Dose required to reduce the lowering of the arterial blood pressure caused by PAF by 50%. Parentheses contain 95% confidence limits. <sup>c</sup> Dose required to inhibit PAF-induced mortality by 50% (intravenous route). <sup>d</sup> Dose required to inhibit PAF-induced mortality by 50% (oral administration). <sup>e</sup> Compounds were purified by chromatography and triturated with ether. <sup>f</sup> Satisfactory elemental analyses ( $\pm 0.4\%$ ) were obtained for C, H, and N for all compounds except for the following products (compd, element, calcd (found value)): 7 N, 15.93 (15.34).

# Scheme 1<sup>a</sup>

 $^a$  (a) BOC<sub>2</sub>O, CHCl<sub>3</sub>, room temperature, 18 h; (b) 4-chloro-3-nitropyridine, Et<sub>3</sub>N, CH<sub>3</sub>Cl, reflux for 18 h, 64% (two steps); (c) H<sub>2</sub>, 10% Pd/C, MeOH, 18 h; (d) Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>, pyridine/H<sub>2</sub>O, room temperature, 18 h; (e) ethyl acetimidate hydrochloride, EtOH, reflux, 18 h, 62% (two steps); (f) 6.5 N HCl<sub>g</sub>/dioxane, MeOH, room temperature, 18 h, 78%; (g) R¹COOH, DCC, HOBT, DMF, room temperature, 18 h; (h) Ph(CH<sub>3</sub>)<sub>2</sub>CCH<sub>2</sub>SO<sub>2</sub>Cl, Et<sub>3</sub>N, CHCl<sub>3</sub>, room temperature, 18 h; (i) R²COOH, (PhO)<sub>3</sub>, Et<sub>3</sub>N, benzene, 90 °C, 2 h and then 8 was added, 90 °C, 18 h; (j) R²NHCOOPh, pyr, 130 °C, 18 h; (k) Ph(C<sub>3</sub>H<sub>4</sub>)CH<sub>2</sub>OCOOPh, pyr, 130 °C, 18 h.

## Chemistry

All the (piperidylmethyl)imidazopyridines were prepared from a common intermediate, amine **8**, which was synthesized according to Scheme 1. BOC-protection of the secondary amine of 4-(aminomethyl)piperidine fol-

lowed by alkylation of the primary amine with 4-chloro-3-nitropyridine<sup>5</sup> gave nitropyridine **10**. This compound was reduced with  $Na_2S_2O_4$  in pyridine/ $H_2O^6$  or alternatively hydrogenated over 10% Pd/C to diamine **11**, which was cyclized to the imidazopyridine **12** using

ethyl acetimidate hydrochloride. Deprotection of the secondary amine of 12 afforded amine 8.

Amide derivatives **13–15**, **17–25**, **32**, **36**, and **37** were obtained by reacting **8** with the corresponding carboxylic acid (R<sup>1</sup>COOH) via DCC coupling in the presence of HOBT. Ureas **26–29**, **31**, and **33–35** were obtained by two methods: either by reaction of **8** with the isocyanate generated through treatment of the corresponding R<sup>2</sup>-COOH with diphenyl phosphorazidate or by coupling 8 with the appropiate phenyl carbamate (R<sup>2</sup>NHCOOPh) in pyridine heating at reflux. Similar conditions were used for the reaction of  ${\bf 8}$  with  $Ph(C_3H_4)CH_2OCOPh$  to give **30**. Sulfonamide **16** was prepared by reaction of **9** with Ph(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>SO<sub>2</sub>Cl.<sup>7</sup> The carboxylic acids are commercially available or were prepared by known methods. 8 Compounds 6 and 7 were obtained from 1-(4piperidyl)-1*H*-2-methylimidazo[4,5-*c*]pyridine. This intermediate was synthesized as described for amine 8 starting from 1-(tert-butoxycarbonyl)-4-aminopiperidine.

#### **Results and Discussion**

Several different assays were used to evaluate the potency of the PAF antagonists described here: the in vitro PAF-induced platelet aggregation assay,9 the in vivo PAF-induced hypotension test in normotensive rats, 10 and the PAF-induced mortality test in mice 11 (po and iv administration). Along with the standard reference compound WEB-2086, we used BB-882, also an imidazopyridine derivative and one of the most advanced PAF antagonists presently in clinical trials. On the basis of the general structure-activity relationship (SAR) trends found for the acyl substituent in our earlier series, five groups of compounds were tested, diaryl- or alkylarylpropanoyl derivatives **13–22**, their 3-hydroxy-substituted analogues 23-25, the urea and carbamate derivatives 26-30, and amino acid derivatives **31–37** (Table 2). Diphenyl- or alkylphenylpropanoyl derivatives 13-15 and 19 gave potent activities in the three tests involving the in vitro assay and iv in vivo tests, 3,3-diphenylpropanoyl derivative **19** being the most active. As we found that unsubstituted 3-phenvlpropanoyl derivative 17 showed a significant loss of activity, we infer that branched substitution on the acyl moiety seems to be important for activity. Introduction of a methoxy group at the 2-position of the aromatic ring, as in compound **18**, also enhanced potency. Considering unsaturated derivatives 20 and 21, we found that the ethyl derivative 21 was more favored and comparable to its ether analogue 22. While few compounds (13-15 and 21) showed modest activities in the oral test, 3,3-diphenylpropanoyl derivative 19 was found to show the highest potency comparable to reference compound BB-882. Interestingly, replacing the amide group in compound 15 with a sulfonamide moiety (compound 16) was detrimental to oral activity.

We next introduced a polar group with the ability to establish hydrogen bonding in the 3,3-diaryl- or 3,3arylalkylpropanoyl chain, a strategy that had increased bioavailability in previous series. Among compounds 23–25, which carry a hydroxy group on position 3, 24 and 25 were only slightly less active than 19 in the in vitro and iv in vivo tests and compounds 23 and 25 proved to be 3 times more active than either 19 or BB-882 in the oral test. Replacing the amide with urea or carbamate was in general well tolerated (compounds

**26–30**) except in the oral test, where only cyclopropyl derivative 29 showed acceptable activity. It was interesting to note that the enantiomer of 27, 28, was significally less active. Taking into account that we found activity in urea derivatives, we tried compounds with the urea function as the linker between the piperidine and the leucine ethyl ester, moiety present in BB-882 (see compound 31), in comparison with ethyl phenylglycines 33 and 34 and the racemic carba analogue 32. Good results were obtained for 31, 33, and its ether analogue **35** in the first three screening tests, but only ether 35 showed modest activity in the oral test.  $\alpha$ -Amino amide **36**, which had the advantage of lacking chiral centers, showed diminished activity that again increased in the presence of a 2-methoxy substituent in the aromatic ring (see compound 37); however, this compound proved orally inactive.

To facilitate the final selection of a lead structure from the most interesting compounds that have emerged from this work (15, 19, 24, 25, 27, and 37), duration of activity studies in two animal species were carried out (see Table 3). All the new compounds showed better results than our previous leader 1 showing that the isosteric replacement of the (cyanomethyl)pyridine moiety with 2-methylimidazo[4,5-c]pyridine produced a substantial increase in the duration of activity. Compound **19** showed the longest duration of activity. It gave 100% protection against PAF-induced mortality in mice 7 h after iv administration of a single dose of 1 mg/kg and also provided 100% inhibition of PAFinduced aggregation in dog whole blood<sup>12</sup> 6 h after iv administration of the same dose.

In conclusion, we have found that replacement of the (cyanomethyl)pyridine moiety present in our previous series with 2-methylimidazo[4,5-c]pyridine results in compounds of greater PAF antagonist activity. Among the different types of acyl radicals tested, the 3,3diphenylpropanoyl substituent of 19 was found to confer the highest potency and the longest duration of activity. The lead structure 19 has been selected for in-depth pharmacological evaluation.

## **Experimental Section**

**Chemistry.** Melting points were determined with a Mettler FP 80 central processor melting-point apparatus and are uncorrected. <sup>1</sup>H NMR (80 MHz) and <sup>13</sup>C NMR (20.1 MHz) spectra were recorded on a Brücker AC80 spectrometer, and <sup>1</sup>H NMR (250 MHz) spectra were recorded on an AVANCE DPX spectrometer and are reported in ppm on the  $\delta$  scale from the indicated reference. Mass spectra were measured on an HP-5988 quadrupole mass spectrometer. Combustion analyses were performed with a Carlo Erba 1106 analyzer. Liquid chromatography was performed with a forced flow (flash chromatography) of the indicated solvent system on SDS silica gel chromagel 60 a C.C. (230-400 mesh). Unless otherwise specified, all nonaqueous reactions were conducted under a rigorously dried argon atmosphere, using oven-dried glassware. C-18-PAF acether was synthesized from (S)-batyl alcohol13 following a published procedure.14

1-(tert-Butoxycarbonyl)-4-(aminomethyl)piperidine (9). To a cooled (0 °C) solution of 4-(aminomethyl)piperidine (150 g, 1.3 mol) in CHCl<sub>3</sub> (800 mL) was added a solution of BOC<sub>2</sub>O (147 g, 0.65 mol) in CHCl<sub>3</sub> (500 mL), and the mixture was stirred for 18 h. After adding H<sub>2</sub>O (400 mL), the organic phase was separated, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated. The resulting material was used directly in the next step: <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 4.11 (br d, J = 13.4 Hz, 2H), 2.69 (m, 2H), 2.58 (d, J = 5.5 Hz, 2H), 1.45 (s, 9H), 1.8-0.8 (m, 7H).

**Table 2.** PAF Antagonist Activities of Compounds **13–37** 

R-N	) N N N	PAF-induced platelet aggregation IC50, <sup>a</sup> µM	PAF-induced hypotension ID50, <sup>b</sup> mg/kg iv	PAF-induced mortality ID50, <sup>c</sup> mg/kg iv	PAF-induced mortality ID50, <sup>d</sup> mg/kg po	mp, <sup>o</sup> C (cryst solvent) <sup>e</sup>	formula <i>f</i>
comp 13	R	0.012	0.033	0.001-0.003	0.3-1	132-135	C23H28N4O.H2O
13	Y II Ph O	(0.007-0.019)	(0.021-0.052)	0.001-0.003	0.3-1	132-133	C2311281140.1120
14	$\underbrace{\hspace{1cm}}_{Ph} \underbrace{\hspace{1cm}}_{O}$	0.011 (0.0095-0.012)	0.0079 (0.005-0.012)	0.003-0.01	0.5-1	110-115	C25H32N4O.3/4H2O
15	Ph O	0.0067 (0.0058-0.0077)	0.012 (0.010-0.14)	0.0027 (0.0018-0.0042)	0.3-1	147-148	C <sub>24</sub> H <sub>30</sub> N <sub>4</sub> O.1/2H <sub>2</sub> O
16	Ph O'S O	0.0062 (0.0051-0.0075)	0.087 (0.054-0.014)	50% inh at 0.01 mg/kg	>1	165-166 (A)	C <sub>23</sub> H <sub>30</sub> N <sub>4</sub> SO <sub>2</sub>
17	Ph	0.060 (0.054-0.065)	>5	0.013 (0.011-0.015)	>1	53-57	C <sub>22</sub> H <sub>26</sub> N <sub>4</sub> O.3/4H <sub>2</sub> O
18	2-MeOPh O	0.015 (0.014-0.007)	0.045 (0.028-0.0072)	0.0033 (0.0028-0.0039)	>1	54-56	C <sub>23</sub> H <sub>28</sub> N <sub>4</sub> O <sub>2</sub> .1/2H <sub>2</sub> O
19	Ph Ph O	0.0076 (0.006-0.011)	0.0086 (0.0063-0.012)	0.0008 (0.0006-0.0010)	0.092 (0.060-0.140)	207-208 (B)	C <sub>28</sub> H <sub>30</sub> N <sub>4</sub> O
20	Ph Ph O	0.24 (0.17-0.35)	0.011 (0.008-0.014)	0.0037 (0.0020-0.0069)	>1	86-91	C <sub>28</sub> H <sub>28</sub> N <sub>4</sub> O.1/4H <sub>2</sub> O
21	Ph	0.013 (0.011-0.017)	0.0054 (0.002-0.014)	50% inh at 0.03 mg/kg	0.3-1	59-62	C <sub>24</sub> H <sub>28</sub> N <sub>4</sub> O.1/2H <sub>2</sub> O
22	Ph	0.018 (0.011-0.029)	0.022 (0.015-0.032)	0.0079 (0.0031-0.020)	>1	63-67	C24H28N4O2.H2O
23	Ph HO Ph O	0.64 (0.50-0.85)	0.013 (0.010-0.016)	0.01-0.03	0.02-0.03	210-211	C <sub>28</sub> H <sub>30</sub> N <sub>4</sub> O <sub>2</sub> .1/2H <sub>2</sub> O
24	HO Ph O	0.0069 (0.0035-0.013)	0.043 (0.034-0.054)	0.0029 (0.0017-0.0050)	0.3-1	199-200	C <sub>23</sub> H <sub>28</sub> N <sub>4</sub> O <sub>2</sub> .1/4H <sub>2</sub> O
25	F <sub>3</sub> C HO Ph O	0.036 (0.028-0.045)	0.013 (0.010-0.015)	0.0029 (0.0019-0.0044)	0.032 (0.023-0.046)	210-211 (C)	C <sub>23</sub> H <sub>25</sub> F <sub>3</sub> N <sub>4</sub> O <sub>2</sub> .1/2H <sub>2</sub> O
26	Ph NH Ph O	0.0081 (0.0062-0.011)	0.14 (0.10-0.18)	>0.03	>1	211-212 (D)	C <sub>27</sub> H <sub>29</sub> N <sub>5</sub> O.1/2H <sub>2</sub> O
27	NH I	0.013 (0.012-0.016)	0.017 (0.012-0.025)	0.0047 (0.0042-0.0052)	0.3-1	80-84	C <sub>22</sub> H <sub>27</sub> N <sub>5</sub> O.1/2H <sub>2</sub> O
28	///NH Ph O	0.041 (0.033-0.051)	0.21 (0.13-0.35)	45% inh at 0.03 mg/kg	>1	79-83	C <sub>22</sub> H <sub>27</sub> N <sub>5</sub> O.1/2H <sub>2</sub> O
29	Ph O	0.017 (0.013-0.024)	0.021 (0.013-0.035)	0.0020 (0.0020-0.0020)	0.2-0.3	227-228	C <sub>23</sub> H <sub>27</sub> N <sub>5</sub> O.1/4H <sub>2</sub> O
30	Ph $O$	0.035 (0.026-0.045)	0.014 (0.011-0.016)	0.012 (0.002-0.070)	>1	138-140	C <sub>24</sub> H <sub>28</sub> N <sub>4</sub> O <sub>2</sub> .1/2H <sub>2</sub> O
31	EtOOC	0.012 (0.008-0.019)	0.030 (0.020-0.046)	0.0027 (0.0012-0.0063)	>1	60-63	C <sub>22</sub> H <sub>33</sub> N <sub>5</sub> O <sub>3</sub> .3/4H <sub>2</sub> O
32	EtOOC Ph O	0.012 (0.0091-0.015)	0.17 (0.09-0.33)	45% inh at 0.03 mg/kg		173-174	C <sub>25</sub> H <sub>30</sub> N <sub>4</sub> O <sub>3</sub> .3/2H <sub>2</sub> O
33	EtOOC NH Ph O	0.016 (0.012-0.020)	0.061 (0.031-0.12)	50% inh at 0.01 mg/kg	>1	153-154	C <sub>24</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub> .1/2H <sub>2</sub> O
34	EtOOC NH Ph O	0.15 (0.12-0.19)	0.069 (0.041-0.11)	0.01-0.02	>1	78-80	C <sub>24</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub> .1/2H <sub>2</sub> O
35	EiO NH Ph O	0.017 (0.013-0.023)	0.019 (0.012-0.029)	0.003-0.01	0.3-1	68-70	C <sub>24</sub> H <sub>31</sub> N <sub>5</sub> O <sub>2</sub> .1/2H <sub>2</sub> O
36	N Ph O	0.023 (0.012-0.044)	0.31 (0.23-0.43)	>0.03	>1	74-78	C <sub>22</sub> H <sub>27</sub> N <sub>5</sub> O.5/4H <sub>2</sub> O
37	2-MeOPh O	0.014 (0.011-0.017)	0.063 (0.043-0.091)	0.0066 (0.0031-0.0143)	>1	59-64	C <sub>23</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub> .1/2H <sub>2</sub> O

Table 2 (Continued)

$\sim \sim N \stackrel{\sim}{\sim} N$	platelet	hypotension	PAF-induced mortality ID50, <sup>c</sup> mg/kg iv	PAF-induced mortality ID50, <sup>d</sup> mg/kg po	mp, °C (cryst solvent) <sup>e</sup>	formula <i>f</i>	
comp R							
WEB-2086 CH <sub>3</sub> -(N-N)							
	0.091 (0.071-0.12)	0.17 (0.12-0.27)	0.084 (0.042-0.167	1.24 ) (0.70-2.21)			
BB-882  O Me N S O	0.0036 (0.0021-0.0063)	0.0037 (0.0024-0.0057)	0.0021 (0.0010-0.004	0.11 0) (0.07-0.19)			

a-d See corresponding footnotes to Table 1. e A, EtOAc/hexane; B, EtOAc/MeOH; C, EtOAc/ether; D, EtOAc. All other compounds were purified by chromatography and triturated with ether. f Satisfactory elemental analyses ( $\pm 0.4\%$ ) were obtained for C, H, and N for all compounds except for the following products (compd, element, calcd (found value)): 17 N, 14.90 (15.79); 31, N, 16.32 (15.64).

Table 3. Duration of Activity of Compounds 15, 19, 24, 25, 27, and 37a

	PAF-induced mortality in mice, inh (%) at 1 mg/kg iv			PAF-induced platelet aggregation, inh (%) at 1 mg/kg iv		
compd	3 h	5 h	7 h	4 h	6 h	
15		88	50	100	100	
19	100	100	100	100	100	
24		80	60	59	43	
25		90	70			
27		56	22	100	77	
37		90	70	100	87	
1	30	15		39		
WEB-2086	55	27	31	100	55	
BB-882	100	95	56	100	100	

<sup>a</sup> Percent protection of PAF-induced mortality in mice and percent inhibition of PAF-induced platelet aggregation in dog whole blood.

4-[[[1-(tert-Butoxycarbonyl)-4-piperidyl]methyl]amino]-**3-nitropyridine (10).** To a mixture of POCl<sub>3</sub> (120 mL) and PCl<sub>5</sub> (112 g) heated at 65 °C was added slowly 4-hydroxy-3nitropyridine (75 g, 0.535 mol). This mixture was stirred at 130 °C for 18 h. The solution was reduced in vacuo to an oil that was redisolved in CHCl<sub>3</sub> (700 mL). After cooling to 0 °C, a solution of 9 (0.65 mol) and Et<sub>3</sub>N (110 mL) in CHCl<sub>3</sub> (500 mL) was added and the mixture heated at reflux for 18 h. It was then evaporated, and the residue was partitioned between  $1\ N$  NaOH and EtOAc. The aqueous phase was reextracted twice with EtOAc, and the combined organic extracts were dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to a total volume of 400 mL. After cooling (-20 °C) overnight, a yellow solid was collected and dried (115 g, 64% for two steps): mp 131-138 °C;1H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 9.20 (s, 1H), 8.30 (d, J = 5.5 Hz, 1H), 8.19 (m, 1H), 6.72 (d, J = 5.5 Hz, 1H), 4.18 (br d, J =13.4 Hz, 2H), 3.26 (t, J = 5.9 Hz, 2H), 2.72 (br t, J = 12.7 Hz, 2H), 1.46 (s, 9H), 1.8-0.8 (m, 5H). Anal. (C<sub>16</sub>H<sub>24</sub>N<sub>4</sub>O<sub>4</sub>) C, H,

3-Amino-4-[[[1-(tert-butoxycarbonyl)-4-piperidyl]meth**yl]amino]pyridine (11).** A mixture of **10** (26.2 g, 0.077 mol) and 10% Pd/C (3.83 g) in MeOH (500 mL) was hydrogenated at atmospheric pressure for 18 h. The insoluble material was removed by filtration and the filtrate evaporated to give 22.9 g (96%) of an oil:  $^1$ H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 7.99 (d, J = 5.5 Hz, 1H), 7.94 (s, 1H), 6.56 (d, J = 5.5 Hz, 1H), 4.20 (br d, J = 13.4 Hz, 2H), 3.19 (m, 5H), 2.81 (br t, J = 12.0 Hz, 2H), 1.58 (s, 9H), 2.1-1 (m, 5H).

Alternatively, 11 was prepared as follows. To a solution of 10 (82.2 g, 0.244 mol) in pyridine (400 mL) was added a solution of  $Na_2S_2O_4$  (170 g, 0.976 mol) in  $H_2O$  (500 mL). The mixture was stirred at room temperature for 24 h and then partitioned between EtOAc (800 mL) and 5 N NaOH (450 mL). The organic phase was separated, dried (MgSO<sub>4</sub>), and evaporated to give a yellow solid (101 g), which still contained pyridine.

1-[[1-(tert-Butoxycarbonyl)-4-piperidyl]methyl]-1H-2**methylimidazo[4,5-***c***]pyridine (12).** A solution of crude product 11 (0.24 mol) and ethyl acetimidate hydrochloride (60.3 g, 0.488 mol) in EtOH (930 mL) was refluxed for 18 h. More ethyl acetimidate hydrochloride (30.15 g, 0.24 mol) was added, and the mixture was heated for an additional 18 h. The solvent was removed in vacuo and the residue partitioned between 2 N NaOH and EtOAc. The organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to an oil that was purified by chromatography on silica gel (CHCl3:MeOH, 4%) to give a creamy solid (49 g, 62% for two steps): 1H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.90 (s, 1H), 8.30 (d, J = 5.5 Hz, 1H), 7.13 (d, J = 5.5 Hz, 1H), 4.10 (br d, J = 13.4 Hz, 2H), 3.90 (d, J = 7.3Hz, 2H), 2.55 (s, 3H), 2.54 (br t, J = 12.7 Hz, 2H), 1.46 (s, 9H), 2.2-1.0 (m, 5H).

1-(4-Piperidylmethyl)-1*H-2*-methylimidazo[4,5-c]pyri**dine (8).** To a cooled solution (0 °C) of **12** (45.1 g, 0.136 mol) in MeOH (470 mL) was added dropwise 6.2 N HCl(g)/dioxane solution (93 mL). After the addition was completed, the cooling bath was removed and the mixture was stirred at room temperature for 18 h. It was then evaporated and the residue partitioned between cooled 2 N NaOH solution and CHCl3. The aqueous phase was reextracted twice with CHCl<sub>3</sub>, and the combined organic extracts were dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give a creamy solid (24.3 g, 78%): mp 128-129 °C; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.96 (s, 1H), 8.35 (d, J = 5.5 Hz, 1H), 7.20 (d, J = 5.5 Hz, 1H), 3.95 (d, J = 7.3 Hz, 2H), 3.06 (br d, J = 12.0 Hz, 2H), 2.61 (s, 3H), 2.51 (br t, J = 12.7 Hz, 2H), 2.2–1.0 (m, 6H). Anal.  $(C_{13}H_{18}N_4\cdot {}^{1}/_{4}H_2O)$ .

1-[[1-[(2-Methyl-2-phenylpropyl)sulfonyl]-4-piperidyl]methyl]-1H-2-methylimidazo[4,5-c]pyridine (16). To a solution of amine 8 (1 g, 4 mmol) and Et<sub>3</sub>N (0.6 mL) in CHCl<sub>3</sub> (20 mL) was added 2-methyl-2-phenylpropanesulfonyl chloride<sup>7</sup> (2.32 g, 10 mmol), and the mixture was stirred at room temperature overnight. The resulting solution was diluted with CHCl3, washed with 0.5 N NaOH, dried, and concentrated. The residue was purified by chromatography on silica gel (CHCl3:MeOH, 5%) to afford a white solid which was recrystallized from EtOAc (0.4 g, 25%): mp 165-166 °C; ¹H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.98 (s, 1H), 8.38 (d, J = 5.5Hz, 1H), 7.32 (m, 5H), 7.17 (d, J = 5.5 Hz, 1H), 3.94 (d, J =6.9 Hz, 2H), 3.65 (br d, J = 12.0 Hz, 2H), 3.16 (s, 2H), 2.60 (s, 2H)3H), 2.18 (br t, J = 12.0 Hz, 2H), 1.58 (s, 6H), 1.48 (m, 5H). Anal. (C23H30N4O2S) C, H, N.

1-[[1-(3,3-Diphenylpropanoyl)-4-piperidyl]methyl]-1H-**2-methylimidazo**[4,5-*c*]**pyridine** (19). To a cooled solution (0 °C) of **8** (19.9 g, 0.083 mol), 3,3-diphenylpropanoic acid (18.7 g, 0.083 mol), and HOBT (11.18 g, 0.083 mol) in DMF (300 mL) was added DCC (17.2 g, 0.083 mol). After the solution was stirred at room temperature for 18 h, the solid material was separated and the filtrate concentrated. The residue was partitioned between NaHCO<sub>3</sub> solution and CHCl<sub>3</sub>. The organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to an oil that was purified by chromatography on silica gel (CHCl<sub>3</sub>:MeOH, 4%) to give a white solid (35 g). Recrystallization from EtOAc afforded a white solid (23.4 g, 68%): mp 207-208 °C; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.90 (s, 1H), 8.45 (d, J = 5.5 Hz, 1H), 7.23 (m, 11H), 4.67 (t, J = 7.4 Hz, 1H), 4.65 (m, 1H), 3.85 (t, J = 7.0 Hz, 2H), 3.83 (m, 1H), 3.07 (q of d, J = 13, 8.1 Hz, 2H), 2.75 (br t, J = 11.8 Hz, 1H), 2.62 (s, 3H), 2.37 (br t, J =11.8 Hz, 1H), 1.98 (m, 1H), 1.58 (br d, J = 11.8 Hz, 1H), 1.41 (br d, J = 11.8 Hz, 1H), 1.08 (q of d, J = 11, 0.4 Hz, 1H), 0.66 (q of d, J = 11, 0.4 Hz, 1H); <sup>13</sup>C NMR (20.15 MHz, CDCl<sub>3</sub>)  $\delta$ (TMS) 169.25, 153.17, 143.78, 143.37, 140.85, 140.71, 139.84, 138.97, 127.93, 127.58, 127.19, 125.87, 104.74, 48.62, 47.11, 44.90, 40.88, 38.00, 36.32, 29.66, 29.03, 13.37. Anal.  $(C_{28}H_{30}N_4O)$  C, H, N.

1-[[1-[[(1-Phenylcyclopropyl)amino]carbonyl]-4-piperidyl]methyl]-1*H*-2-methylimidazo[4,5-c]pyridine (29). Diphenyl phosphorazidate (2.14 mL, 0.01 mol) was added dropwise to a solution of 1-phenyl-1-cyclopropanecarboxylic acid (1.6 g, 0.01 mol) and Et<sub>3</sub>N (1.14 mL) in benzene (40 mL). The mixture was heated at 90 °C for 2 h. Amine 8 (1.6 g, 0.0068 mol) was then added, and the mixture was heated at 90 °C for 18 h more. After cooling the mixture was partitioned between EtOAc and 1 N NaOH solution. The aqueous phase was reextracted twice with EtOAc, and the combined organic phases were dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated. The crude product was purified by chromatography on silica gel (CHCl3: MeOH, 10%) to give a white solid (1.2 g, 46%): mp 227-228 °C; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.98 (s, 1H), 8.38 (d, J = 5.5 Hz, 1H), 7.20 (m, 6H), 5.44 (s, 1H), 3.97 (d, J = 7.3 Hz, 2H), 3.95 (m, 2H), 2.62 (s, 3H), 2.60 (br t, J = 12.7 Hz, 2H), 2.2-1.0 (m, 5H), 1.22 (s, 4H). Anal (C<sub>23</sub>H<sub>27</sub>N<sub>5</sub>O·¹/<sub>4</sub>H<sub>2</sub>O) C, H,

1-[[1-[[(1-Pheny-1-cyclopropyl)methoxy]carbonyl]-4piperidyl]methyl]-1*H*-2-methylimidazo[4,5-c]pyridine (30). Amine 8 (0.72 g, 0.0031 mol) was added to a solution of phenyl [(1-phenyl-1-cyclopropyl)methyl]carbonate (prepared from reaction of (1-phenyl-1-cyclopropyl)methanol with phenyl chloroformate; 1.1 g, 0.0041 mol) in pyridine (30 mL). The mixture was refluxed for 18 h. After removal of the solvent in vacuo, the residue was partitioned between CHCl<sub>3</sub> and 1 N NaOH. The aqueous phase was reextracted twice with CHCl<sub>3</sub>, and the combined organic phases were dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated. The crude product was purified by chromatography on silica gel (CHCl<sub>3</sub>:MeOH:NH<sub>3</sub>, 60:2:0.2) to give a white solid (0.3 g, 25%): mp 138–140 °C; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.97 (s, 1H), 8.37 (d, J = 5.5 Hz, 1H), 7.27 (s, 5H), 7.19 (d, J= 5.5 Hz, 1H), 4.16 (s, 2H), 4.15 (br d, J = 13.6 Hz, 2H), 3.94 (d, J = 7.21 Hz, 2H), 2.60 (s, 3H), 2.60 (m, 2H), 2.3-0.8 (m, 5H), 0.92 (s, 4H). Anal. (C<sub>24</sub>H<sub>28</sub>N<sub>4</sub>O<sub>2</sub>·¹/<sub>2</sub>H<sub>2</sub>O) C, H, N.

Biological Methods: Inhibition of Platelet Aggregation in Vitro. Platelet aggregation studies were done by the method of Born.9 Blood was collected in 3.8% sodium citrate (1 vol/9 vol of blood) by cardiac puncture from male New Zealand rabbits (2-2.5 kg body weight). Platelet-rich plasma (PRP) was prepared by centrifuging the blood at 250g for 10 min at 4 °C. The PRP was diluted with platelet-poor plasma obtained by further centrifuging at 3000g for 10 min. The platelet number was adjusted to  $3.5 \times 10^5$  cells/mm<sup>3</sup>. Platelet aggregation was induced by C-18-PAF (1.5  $\times$  10<sup>-8</sup> M) and measured with a dual-channel aggregometer Chrono-log 560 instrument. Activity is expressed as the IC<sub>50</sub> value, i.e., the concentration required to inhibit platelet aggregatory response by 50%. The values shown in the tables were calculated by linear regression from a single experimental curve with no less than four data points, each point being the mean of the percentage inhibition at a given concentration obtained from one to three independent experiments.

Inhibition of PAF-Induced Hypotension in Normotensive Rats. 10,1 Male Sprague—Dawley rats, weighing 180— 220 g, were anesthetized with sodium pentobarbital (50 mg/ kg ip). Blood pressure was recorded from the left carotid artery using a Statham pressure transducer coupled to a Beckman R611 recorder. Right and left femoral veins were catheterized to inject the test compound and PAF (0.5  $\mu$ g/kg). Test compounds were administered by intravenous injection (1 mL/kg, dissolved in saline) 3 min before PAF injection. Blood pressure was monitored, and percentage inhibition of PAF-induced hypotension with respect to controls was calculated. The results are expressed as ID<sub>50</sub> values, i.e., the dose

of test compound required to inhibit hypotension by 50%, or as percentage inhibition at a given dose of test compound. The ID<sub>50</sub> values were calculated by linear regression from a single experimental curve with no less than four points, each point being the mean of the percentage inhibition at a given dose obtained from two or more independent experiments.

**Inhibition of PAF-Induced Mortality in Mice.** <sup>11</sup> Groups of 10 male Swiss mice weighing 22–26 g were used; 100  $\mu$ g/ kg C-18-PAF plus 1 mg/kg propanolol was administered through a lateral tail vein 5 min after iv or 60 min after po administration of the test compounds (10 mL/kg iv or 20 mL/ kg po) or vehicle in control groups (saline iv or 1% Tween 80 po). The animals were observed 2 h after the PAF injection. Following this protocol, we obtained a consistent mortality of 70-100% in the control group. Percentage inhibition of mortality due to treatment in comparison with the control group was calculated. Results are given as ID<sub>50</sub> values, i.e., the dose required to inhibit PAF-induced mortality by 50%. The results were calculated by linear regression from a single experimental curve with at least four data points. For duration of activity studies, test compounds were given at a single dose of 1 mg/kg iv at time 0. C-18-PAF plus propanolol was administered at different times (10-40 mice/time), and percentage inhibition at each time was recorded.

Inhibition of PAF-Induced Platelet Aggregation in **Dog Whole Blood.** Blood was collected by venipuncture from male Beagle dogs (8-12 kg body weight) before and 4 and 6 h after the administration of a single dose of 1 mg/kg iv of the test compounds; 3.8% sodium citrate (1 vol/9 vol of blood) was used as the anticoagulant. Blood was further diluted 1:1 (v/ v) with saline. Platelet aggregation was induced by C-18-PAF  $(7.5 \times 10^{-9} \text{ M})$  and measured by an impedance method  $^{12}$  with a dual-channel aggregometer Chrono-log 560 instrument. Percentage inhibition of platelet aggregation at 4 and 6 h was calculated by comparison with the platelet response in the same animal before the administration of the test compound. Results are the mean percentage inhibition obtained from one

Statistics. Statistical analyses of pharmacological data were done with a standard pharmacology program. 15

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